



## A Facile Immobilization Approach for Perfunctionalised Cyclodextrin onto Silica via the Staudinger Reaction

Li-feng Zhang<sup>a</sup>, Yeang-Chyn Wong<sup>b</sup>, Lei Chen<sup>b</sup>, Chi Bun Ching<sup>a</sup> and Siu-Choon Ng<sup>b\*</sup>

<sup>a</sup> *Department of Chemical & Environmental Engineering, National University of Singapore, Singapore 119260*

<sup>b</sup> *Department of Chemistry, National University of Singapore, Singapore 119260*

Received 27 November 1998; accepted 23 December 1998

**Abstract:** The Staudinger reaction was applied for the first time in the immobilization of mono-(6-azido-6-deoxy)-perfunctionalised cyclodextrins onto the surface of aminised silica gel under mild conditions. The composite materials obtained are applicable as chiral stationary phases for enantioseparation processes.

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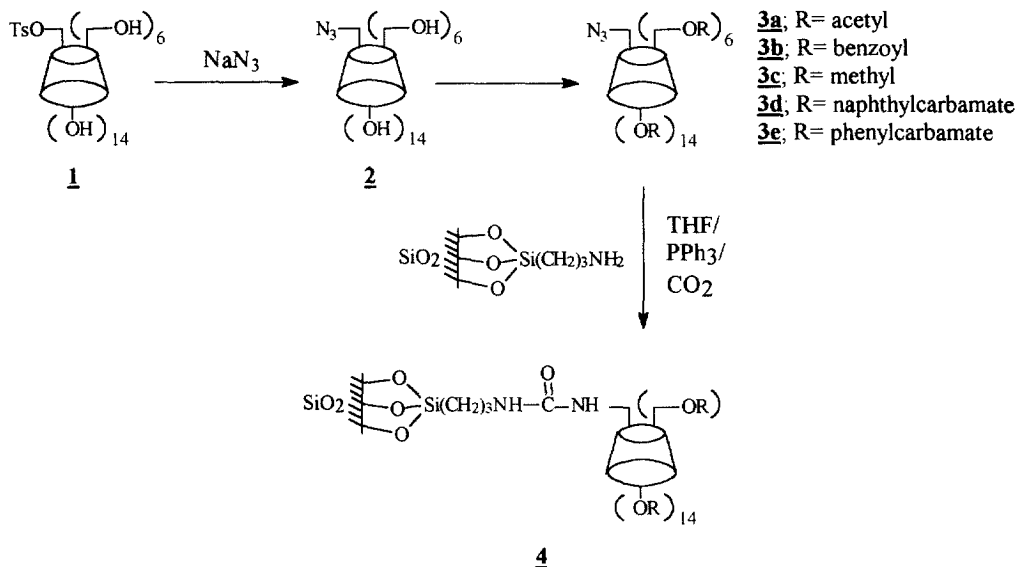
Functionalised cyclodextrins have been attracting burgeoning interests since the demonstration by Armstrong and co-workers<sup>1</sup> that immobilization of these chiral compounds onto inert support materials, such as silica gel afforded chiral stationary phases (CSP) which are amenable to both analytical and preparative scale enantioseparation of a broad range of racemic compounds. Previously, research efforts were directed mainly on chemically anchoring cyclodextrins (CD) onto solid matrices *via* amino linkages<sup>2</sup> and the solvolytically more stable carbamic acid moieties<sup>3</sup> involving post-immobilization derivatisation procedures. In all these approaches, regioselective immobilization of CD cannot be readily accomplished since reaction may occur indiscriminately at the 2, 3 or 6- position of cyclodextrin or result in a complex mixture of multi-anchored CD instead of one with well-defined chemical structure. In addition, subsequent chemical derivatisation of the immobilized CD also does not readily lead to materials with easily controllable batch-to-batch reproducibility on account of the invariable need of conducting heterogeneous solid-liquid reactions. These factors, in part, may have contributed to the high cost and poor availability of the CD-CSP.

We report herein a facile synthetic approach which effectively overcome the above shortcomings. In this, mono-

\* Email: chmngsc@leonis.nus.edu.sg

(6-azido-6-deoxy)perfunctionalised cyclodextrins were first synthesized, purified and characterized. Immobilization onto the surface of aminized silica gel can then be conveniently carried out on the CD-azido functionality using a Staudinger reaction <sup>4</sup> under extremely mild reaction conditions. Chemical anchoring of the CD moieties onto the support were effectively *via* the hydrolytically stable urethane linkage.

**Scheme 1. Synthetic route to perfunctionalised cyclodextrin immobilised silica gel**



The synthetic route used is depicted in Scheme 1 using the readily available mono-6-deoxy-6-(p-tolylsulphonyl)cyclodextrin **1** <sup>5</sup> as starting material. Treatment of **1** with an excess of sodium azide in DMF at 95 °C for 10 hrs afforded mono-(6-azido-6-deoxy)-cyclodextrin **2** after purification by complex formation <sup>6</sup> with 1,1,2,2-tetrachloroethane in *ca* 90% yield. Thereafter, complete functionalisation of the remaining hydroxyl groups can be readily effected under a variety conditions <sup>7</sup> which afforded a range of mono-(6-azido-6-deoxy)-perfunctionalised-cyclodextrins **3** in good yields and high purities.

Aminized silica gel was prepared by stirring silica gel (dried overnight in 0.1 mmHg vacuum at 180 °C) with  $\gamma$ -aminopropyltriethoxysilane in accordance with a literature procedure <sup>8</sup>. Staudinger reaction of the CD-derivatives **3** were initiated by passage of CO<sub>2</sub> into a THF suspension of the aminized silica gel. At this stage, an infrared of a sample of the reacting silica gel depicted vibrational bands at 1560 cm<sup>-1</sup> ascribed to the formation of the propylammonium propylcarbamate on the surface of silica, an expected intermediate in the reaction. Completion of the immobilization process was then effected by addition of a solution mixture of triphenylphosphine and the respective derivatives **3**. The crude CSPs were purified by soxhlet extraction with acetone to remove the triphenylphosphine oxide and any unreacted cyclodextrin by-products.

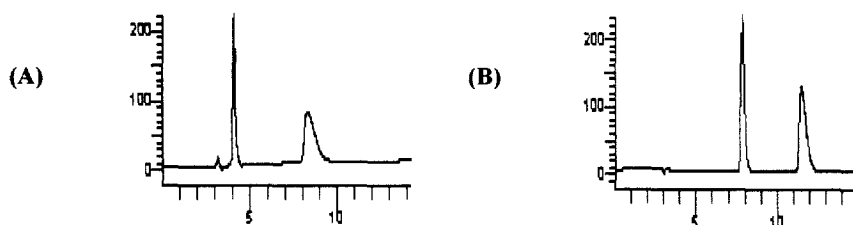
The obtainment of CSPs **4** were evident from the weak but characteristic FT-IR vibrational bands particularly in

the 1400-1800  $\text{cm}^{-1}$  region, reminiscent of those present in the precursor compounds **3**. In addition, the significantly higher carbon contents in the elemental analyses for CSPs **4** as of the presence of determined surface concentration of CD-derivatives (Table 1) further corroborated the success of our immobilization approach. Since we have utilized only purified and characterized precursor compounds **3** and since there is only one azido functional group in every cyclodextrin molecule from which the Staudinger reaction can occur, the current procedure therefore afforded structurally well-defined CSPs.

**Table 1** Charaterisation of the perfunctionalised cyclodextrin immobilised silica gel.

List of Bonded Sorbents	C%	H%	N%	Surface Concentration ( $\mu\text{mol m}^{-2}$ )
Aminised silica gel	5.43	1.82	2.11	---
Permethyl- $\beta$ -CD immobilised silica gel	15.49	3.07	1.52	0.80
Peracetyl- $\beta$ -CD immobilised silica gel	16.70	2.98	1.56	0.67
Perbenzoyl- $\beta$ -CD immobilised silica gel	17.25	2.40	1.43	0.32
Perphenylcarbamate- $\beta$ -CD immobilised silica gel	17.42	2.15	2.56	0.32
Pernaphthylcarbamate- $\beta$ -CD immobilised silica gel	16.17	2.07	2.45	0.20

The respective CSPs when packed into stainless steel HPLC columns, depicted excellent enantioseparation of a board range of structurally diverse racemic compounds and drugs (representative chromatogram depicted in Figure 1). These analytical results will be reported elsewhere.



**Figure 1.** HPLC separation of the enantiomers on stainless column ( $\phi 4.6 \times 250$  mm) packed with perphenylcarbamate- $\beta$ -CD immobilised silica gel. Flow rate: 1ml/ min. UV detector:  $\lambda = 230\text{nm}$ . (A) Atropine, mobile phase: Buffer(1% TEA aqueous adjusted with HOAc, PH= 4.65)/MeOH =65/35; (B) 1-(p-chlorophenyl) ethanol, mobile phase: Hexane/ IPA= 90/10.

## Acknowledgement

Funding from NUS and NSTB-Singapore in support of this project is gratefully acknowledged. Y.C. Wong and L.Chen thanks the NUS for the award of a research scholarship.

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- 3a** was prepared in (CH<sub>3</sub>CO)<sub>2</sub>O/ pyridine in yield of 82%, m.p. 133-135 °C; [α]<sub>D</sub> +120.92 ° (C<sub>1.0</sub>, CHCl<sub>3</sub>); IR (cm<sup>-1</sup>): 2966 (C-H str), 2111 (-N<sub>3</sub> str), 1750 (C=O str), 1052 (C-O str); <sup>1</sup>H NMR (CDCl<sub>3</sub>, TMS) δ (ppm): 5.33-5.26 (m, 7H), 5.13-5.01 (m, 7H), 4.85-4.77 (m, 7H), 4.58-4.54 (d, J= 12.3Hz, 7H), 4.32-4.18 (m, 7H), 4.15-4.10 (m, 7H), 3.80-3.66 (m, 7H), 2.16-2.02 (multi s, 60H); Anal. Calcd. for C<sub>82</sub>H<sub>109</sub>O<sub>54</sub>N<sub>3</sub>: C 49.23%, H 5.49%, N 2.10%; found: C 49.14%, H 5.64%, N 1.95%. **3b** was prepared in PhCOCl/ pyridine in the yield of 66%, m.p. 96-99°C; [α]<sub>D</sub> +24.65 ° (C<sub>1.0</sub>, CHCl<sub>3</sub>); IR (cm<sup>-1</sup>): 3071 (arom C-H str), 2105 (-N<sub>3</sub> str), 1729 (C=O str), 1603, 1449 (C=C arom ring str), 1273, 1097 (C-O-C str), 704 (C-H arom); <sup>1</sup>H NMR (CDCl<sub>3</sub>, TMS) δ (ppm): 8.24-8.12 (m, 12H), 7.75-7.24 (m, 46H), 7.23-7.13 (m, 14H), 7.09-6.80 (m, 28H), 6.10-5.88 (m, 7H), 5.75-5.42 (m, 7H), 5.20-4.50 (m, 28H), 4.24-3.95 (m, 7H); Anal. Calcd. for C<sub>182</sub>H<sub>149</sub>N<sub>3</sub>O<sub>54</sub>: C 67.42%, H 4.63%, N 1.30%; found: C 66.91%, H 4.51%, N 1.25%. **3c** was prepared in CH<sub>3</sub>I/DMF/NaH in the yield of 82%, m.p. 95-99 °C; [α]<sub>D</sub> +157.00 ° (C<sub>1.0</sub>, CHCl<sub>3</sub>); IR (cm<sup>-1</sup>): 2929 (C-H); 2103 (-N<sub>3</sub> str), 1034 (C-O str); <sup>1</sup>H NMR (CDCl<sub>3</sub>, TMS) δ (ppm): 5.18-5.08 (m, 7H), 3.95-3.38 (m, 95H), 3.23-3.15 (m, 7H); Anal. Calcd. for C<sub>62</sub>H<sub>109</sub>N<sub>3</sub>O<sub>34</sub>: C 51.70%, H 7.62%, N 2.92%; found: C 51.43%, H 7.00%, N 2.76%. **3d** was prepared in naphthyl isocyanate/ pyridine in the yield of 55%, m.p. 192-199 °C; [α]<sub>D</sub> +89.56 ° (C<sub>1.0</sub>, CHCl<sub>3</sub>); IR (cm<sup>-1</sup>): 3401, 3308 (N-H str), 2110 (-N<sub>3</sub> str), 1738 (C=O str), 1549, 1505 (arom C=C str), 1215, 1043 (C-O-C str), 772 (C-H arom); <sup>1</sup>H NMR (CDCl<sub>3</sub>, TMS) δ (ppm): 6.20-8.40 (m, 160H), 2.50-5.75 (m, 49H); Anal. Calcd. for C<sub>262</sub>H<sub>209</sub>N<sub>23</sub>O<sub>54</sub>·4H<sub>2</sub>O: C 68.17%, H 4.74%, N 6.98%; found: C 68.45%, H 4.93%, N 6.62%. **3e** was prepared in phenyl isocyanate/ pyridine in the yield of 57%, m.p. 220-227 °C; [α]<sub>D</sub> +33.40 ° (C<sub>1.0</sub>, CHCl<sub>3</sub>); IR (cm<sup>-1</sup>): 3393, 3313 (N-H str); 2102 (N<sub>3</sub> str), 1732 (C=O str), 1604, 1537, 1443 (arom C=C str), 1221, 1053 (C-O-C str), 750 (C-H arom); <sup>1</sup>H NMR (CDCl<sub>3</sub>, TMS) δ (ppm): 7.48-6.55 (m, 120H); 5.65-5.50 (m, 7H), 5.32-5.08 (m, 14H), 4.82-4.25 (m, 21H), 4.05-3.90 (m, 7H); Anal. Calcd. for C<sub>182</sub>H<sub>169</sub>N<sub>23</sub>O<sub>54</sub>·8H<sub>2</sub>O: C 59.29%, H 5.06%, N 8.73 %; found: C 59.59 %, H 5.34 %, N 8.48 %.
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